## ISOLATION AND IDENTIFICATION OF A SUPPLEMENTARY COMPONENT

OF THE SEX PHEROMONE OF Mamestra brassicae FEMALES

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Using gas-liquid chromatography and mass spectrometry, from an extract of the sex pheromone of females of the cabbage moth <u>Mamestra</u> <u>brassicae</u> L. an additional component of the pheromone has been isolated and identified - heptadeccis-ll-en-l-ol acetate.

It has been previously shown convincingly that the main component of the sex pheromone of females of the cabbage moth <u>Mamestra</u> <u>brassicae</u> L. (Lepidoptera: <u>Noctuidae</u>) is hexadec-cis-11-en-1-ol acetate [1-4]. However, literature information on the biological activity of synthetic hexadec-cis-11-en-1-ol alone under field conditions is contradictory [5-7], and its species specificity is fairly low [8]. We are therefore making a search for additional components in order to increase the activity and selectivity of the synthetic cabbage moth pheromone. A repeat analysis of extracts of the sex pheromone showed the additional presence, in an amount of 1-5% of the main component, of acetates of the saturated tetradecanol and hexadecanol, of hexadec-trans-11-en-1-ol [9], of hexadec-cis-9-en-1-ol acetate [10], and of heptadec-cis-11-en-1-ol acetate [4]. In field experiments, additions of these substances to the main component did not increase its power of attraction [5]. According to [11], 1% of hexadec-cis-11-enal substantially increases the activity of hexadec-cis-11-en-1-ol. The hypothesis has been expressed that the main component has been missed or that geographically remote races of the cabbage moth are not identical.

We have continued a study of the composition of a crude extract of the pheromone glands of cabbage moth females located in the south-western zone of the USSR with the aim of finding additional components.

To reveal the possible presence of heptadec-cis-11-en-1-ol acetate, the crude extract in an amount of 50 female-equivalents was separated by preparative GLC on a column containing the nonpolar liquid phase Apiezon L with the collection of one-minute fractions which were tested electroantennographically by a procedure described previously [12].

Figure 1 shows the responses of the antennae of cabbage moth males to the fractions obtained and a chromatogram of synthetic acetates of tetradec-, hexadec-, and heptadec-cis-11-en-1-ols. The most active fraction (11-12) contained the main component of the pheromone - hexadec-cis-11-enyl acetate (component 1). A small response was obtained to fraction 17-18, corresponding in retention time to heptadec-cis-11-enyl acetate.

Fraction 17-18 was isolated preparatively in an amount of 1000 female-equivalents. The main component present in this fraction (component 2) coincided in its retention time with synthetic heptadec-cis-11-en-1-ol acetate on columns with phases of different polarities: SE-30, Apiezon L, XE-60, and diethyleneglycol succinate.

The product of the alkaline hydrolysis of component 2 gave GLC peaks with the same retention indices as those of synthetic heptadec-cis-ll-en-l-ol and its acetate with phases of different polarities:

| Substance   | Retention                    | index, I                     |
|---|------------------------------|------------------------------|
| Bubscance   | S <i>E-30</i>                | XE-60                        |
| cis-11-C <sub>17</sub> OAc<br>cis-11-C <sub>17</sub> OH<br>Component 2<br>Product of the hydrolysis<br>of component 2 | 2075<br>1966<br>2075<br>1966 | 2310<br>2267<br>2310<br>2267 |

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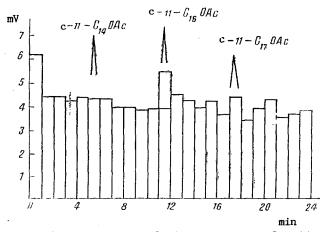


Fig. 1. Responses of the antennae of cabbage moth males to one-minute fractions of a crude extract and a chromatogram of the synthetic acetates of tetradec-, hexadec-, and heptadeccis-11-en-1-ols. Column containing Apiezon L, 200°C, rate of flow of nitrogen 40 ml/min.

The position of the double bond in component 2 was determined by comparing the retention times of the acetoxy aldehydes obtained on the micro-ozonolysis of synthetic acetates with the authentic structures of the hexadec-cis-9-, cis-10, cis-11, and cis-12-en-1-ols and the products of the micro-ozonolysis of component 2.

The product of the micro-ozonolysis of component 2 coincided in its retention time in a column containing the stationary phase SE-30 with 11-acetoxyendecanal — the product of the micro-ozonolysis of hexadec-cis-11-enyl acetate, i.e., the double bond was present in position 11:

| Compound  | Ozonolysis product   | Retention time,<br>min          |
|---|--|---------------------------------|
| $CH_{3}(CH_{2})_{5}CH = CH(CH_{2})_{8}OAc$<br>$CH_{3}(CH_{2})_{4}CH = CH(CH_{2})_{9}OAc$<br>$CH_{3}(CH_{2})_{3}CH = CH(CH_{2})_{10}OAc$<br>$CH_{3}(CH_{2})_{2}CH = CH(CH_{2})_{11}OAc$<br>Component 2 | OHC(CH <sub>2</sub> ) <sub>8</sub> OAc<br>OHC(CH <sub>2</sub> ) <sub>9</sub> OAc<br>OHC(CH <sub>2</sub> ) <sub>10</sub> OAc<br>OHC(CH <sub>2</sub> ) <sub>11</sub> OAc | 2,2<br>3,2<br>4,6<br>6,1<br>4,6 |

The structure of component 2 was confirmed by mass spectrometry. The mass spectrum of the natural sample taken at 20 eV contained a peak with m/z 236 that is characteristic for acetates of monoenic alcohols and corresponds to the ejection of acetic acid from the molecular ion  $[M - AcOH]^+$ , and also the peaks of fragments corresponding to the hydrocarbon breakdown of this ion  $[M - AcOH - nCH_2 = CH_2]^+$ , and  $[M - AcOH - nCH_2]^+$ . The mass spectra of the natural sample and of the synthetic heptadec-cis-ll-enyl acetate (at 20 eV) proved to be practically identical (I, %):

| m z         | Natural sample | Synthetic hexadec-cis-ll-<br>enyl acetate |
|-------------|----------------|---|
| 23 <b>6</b> | 65             | 95  |
| 208         | 40             | 27  |
| 194         | 25             | 20  |
| 180         | 30             | 40  |
| 166         | 30             | 30  |
| 152         | 38             | 63  |
| 138         | 57             | 65  |
| 124         | 75             | 93  |
| 110         | 100            | 100                                       |

Thus, an additional component in an amount of 25% of the main component (according to GLC results) has been found in an extract of the pheromone glands of cabbage moth females, and this is heptadec-cis-11-en-1-ol acetate.

## EXPERIMENTAL

Chromato-mass spectrometry was performed on a MKh-1321 instrument. Separation was carried out on a column 2 m long filled with 3% of OV-1 on the support Gas-Chrom with a grain size of 100/120 mesh.

Gas chromatography was performed on a Chrom-42 instrument using 3 mm  $\times$  2.5 m glass columns filled with 8% of Apiezon L on Chromaton N-AV-HMDS, 80/100 mesh, 3% of SE-20 on Chromaton Super, 80/100 mesh, 5% of XE-60 on Chromaton N-AW-HMDS, 80/100 mesh, and 6% of diethyleneglycol succinate on Chromaton N-AW-HMDS, 80/100 mesh.

The antennograms were recorded by the method and on the apparatus described in [12].

The synthetic samples used in the investigation were obtained via acetylenic alcohols.

<u>Preparation of the Extract</u>. The tips of the abdomens of 2- to 3-day-old females containing the gland producing the pheromone were cut off and extracted with methylene chloride at 5°C for 3-5 days. The extract was filtered, the residue was washed several times with the solvent, and the extract was made up to a definite volume.

<u>Collection of One-Minute Fractions</u>. To record analytical chromatograms and for the preparative collection of one-minute fractions we used an extract from 50 females. The fractions were collected in 1 mm  $\times$  30 cm glass capillaries which were attached to the outlet of the column.

<u>Preparation of a Natural Sample for Mass Spectrometry</u>. The fraction corresponding to the second component of the pheromone was isolated by preparative GLC on a column containing SE-30 from the crude extract obtained from 400 females.

<u>Micro-Ozonolysis of the Pheromone</u>. Micro-ozonolysis was performed by the method of Berozza and Bierl [13]. The ozonolysis products were chromatographed on a 3 mm  $\times$  2.5 m glass column filled wih 3% of SR-30 on Chromaton Super 80/100 mesh.

<u>Hydrolysis of the Pheromone</u>. The second component of the pheromone isolated from 100 females was saponified by being boiled with 2 ml of 10% aqueous caustic soda and 2 ml of ethanol for 25 h. Then the mixture was extracted with ether and the extract was dried with  $Na_2SO_4$  and evaporated and the residue was used in GLC.

## SUMMARY

The acetate of heptadec-cis-11-en-1-ol has been isolated from an extract of the sex pheromone of females of the cabbage moth <u>Mamestra brassicae</u> and identified.

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